

Studies on the reduction and reductive alkylation of amino acid-derived spirocyclic 2,6-dioxopiperazines

Juan A. González-Vera, Pilar Ventosa-Andrés, Joanne Casey,
M. Teresa García-López and Rosario Herranz*

Instituto de Química Médica (CSIC), Juan de la Cierva 3, E-28006 Madrid, Spain

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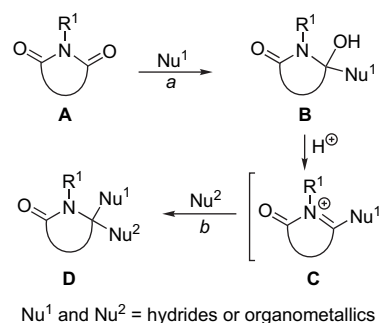
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Abstract—The regio- and diastereoselective reduction and reductive alkylation of 3-spiro-2,6-dioxopiperazines are described via a two-step process, which involves addition of NaBH₄ or Grignard reagents, followed by TFA-mediated dehydration with a second NaBH₄ addition. The results show that the reactivity of 2,6-dioxopiperazines is limited by their steric hindrance and by the volume of the nucleophile, which preferably add to the C₆ carbonylic carbon with complete diastereoselectivity. The diastereoselectivity of the first step is mainly governed by electronic factors, which direct the addition of the nucleophile from the most hindered face, while in the second step, the NaBH₄ attacks from the less crowded face. This second step proceeds with partial or complete racemization.

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1. Introduction

Piperazines and their spiro derivatives are included among the most frequently occurring privileged substructures found in compounds of therapeutic interest.¹ Amongst them, 1,4-disubstituted piperazine derivatives^{1b,d} and 2,5-dioxopiperazines (cyclic dipeptides)^{1c,2} have been the main focus of interest. However, 2,6-dioxopiperazine derivatives have received scarce attention, apart from certain antitumoral bis(2,6-dioxopiperazine) derivatives (topoisomerase II inhibitors),^{2a,3} and some inhibitors of the influenza virus endonuclease, such as flutimide.⁴ From the synthetic point of view, the 2,6-dioxopiperazine system could be considered as a versatile template for the synthesis of diversity of scaffolds, due to the presence of the reactive endocyclic imide group. This group is highly reactive toward nucleophiles,⁵ and is a potential precursor of *N*-acyliminium ions,⁶ which are very useful and versatile reactive species in organic synthesis. Perhaps, one of the most successful and versatile methods for obtaining *N*-acyliminium ion precursors has been the selective addition of hydride or organometallic reagent to one carbonyl group of a cyclic imide.^{5,6} As shown in Scheme 1, most of the described methods for the reduction of cyclic imides (**A**) to lactam derivatives (**D**) involve a two-step process: a first step of addition of a nucleophile (a metal hydride⁷ or an organometallic compound⁸) to give a hydroxylactam (**B**). In the second step, this hydroxylactam in acid media generates an unstable *N*-acyliminium ion intermediate (**C**), which adds a second nucleophile (Et₃SH^{8b-d,9} or an organometallic compound^{7a,c,8a,b}).



Scheme 1. General process of cyclic imide reduction to lactams.

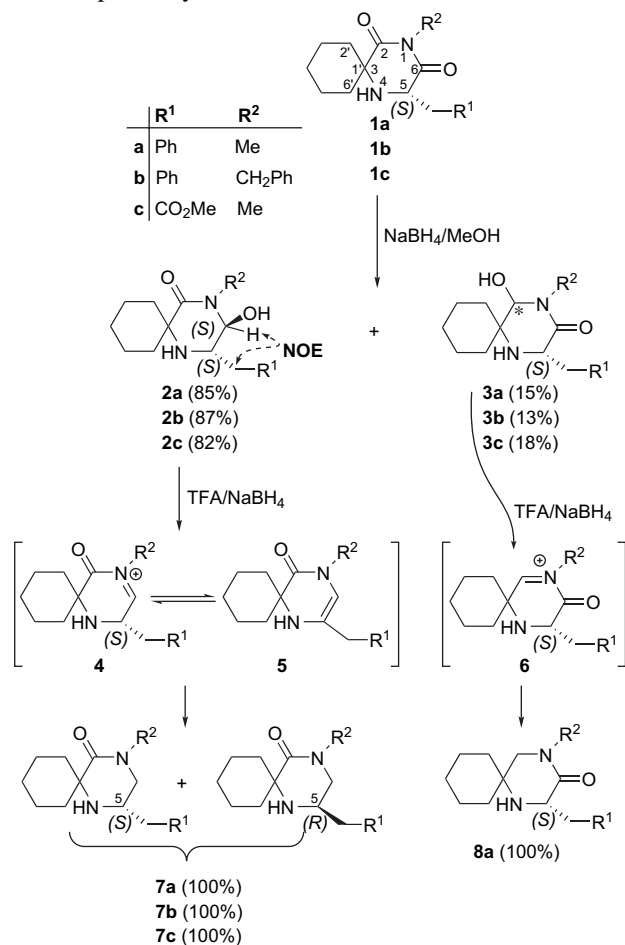
In the course of a project focused on diversity-oriented synthesis, we have recently reported a general method for the synthesis of α -amino acid-derived spirocyclic 2,6-dioxopiperazines from α -quaternary- α -amino nitriles.¹⁰ Now, having in mind the synthetic potential of cyclic imides, we have explored and reported herein our studies on reduction and reductive alkylation of 2,6-dioxopiperazine-3-spirocyclohexane derivatives as a source of diverse and highly substituted spirocyclic piperazines.

2. Results and discussion

Spirocyclic 2,6-dioxopiperazines **1a** and **1b**, derived from L-phenylalanine, and **1c**, derived from L-aspartic acid (Scheme 2),¹⁰ were selected for the studies reported herein. First, we studied the reduction of these dioxopiperazines with NaBH₄. As shown in Scheme 2, treatment of **1a-c** with 3 equiv of NaBH₄ in MeOH at room temperature quantitatively led to a (6:1) mixture of the corresponding

* Corresponding author. Tel.: +34 91 5622900; fax: +34 91 5644853; e-mail: rosario@iqm.csic.es

6-hydroxy- and 2-hydroxypiperazine derivatives **2a–c** and **3a–c**, respectively.



Scheme 2. NaBH₄ mediated reduction of spirocyclic 2,6-dioxopiperazines **1**.

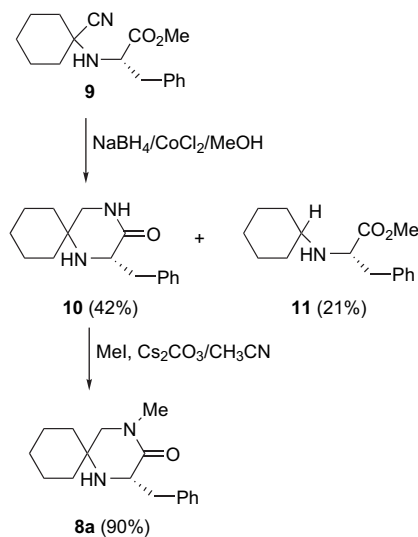
The ¹H NMR spectra of 6-hydroxy-2-oxopiperazines **2a–c** showed the presence of the 6-H proton as a doublet (**2a,b**), coupled to 5-H, or as a triplet (**2c**), coupled to 5-H and 6-OH, at δ 4.46–4.49 ppm, while the spectra of the 2-hydroxy isomers **3a–c** showed the presence of 2-H as a singlet at δ 4.40–4.44 ppm. The ¹³C NMR spectra of these hydroxylactams showed the disappearance of one of the carbonylic carbons and the appearance of a new carbon in the aliphatic carbon region (83–87 ppm). The reduction position was confirmed by the ¹H–¹³C correlations observed in the HMBC spectra. As indicated in **Scheme 2**, the stereochemistry at C₆ in the 6-hydroxy-2-oxopiperazines **2a–c** was assigned as 6*S* based on the NOE effect observed between 6-H and 5-CH₂ protons in their 1D NOESY spectra. These results showed that the nucleophilic attack of NaBH₄ at C₆ was completely diastereoselective, *syn* to the amino acid side chain. As it has been previously proposed for the addition of nucleophiles to cyclic ketones,¹¹ this high preference for the axial addition of the nucleophile on the most hindered face of the dioxopiperazine ring could be due to the higher electronic stabilizing *syn* interaction of the emerging σ* orbital with the C₅–H bond than with the C₅–C bond. The addition of NaBH₄ to C₂ was also completely stereoselective, obtaining only one of the possible diastereoisomers **3a–c**, but, due to the absence of NOE effects on 2-H, we could not assign the configuration at C₂.

In most of the reported precedents of reduction of cyclic imides to lactams,^{8b–d,9} the reduction of the hydroxylactam **B** in the second step of **Scheme 1** is carried out with Et₃SiH, using BF₃·OEt₂ as acid to generate the intermediate *N*-acyliminium ion **C**. However, taking into account our experience in the reduction of 5-hydroxypyrrolidin-2-ones with NaBH₄ in neat TFA,¹² we applied these reagents to the reduction of the major 6-hydroxypiperazines **2a–c** and the minor 2-regioisomer derived from phenylalanine **3a**. This reduction quantitatively yielded the oxopiperazines **7a–c** and **8a**. We also tried to transform directly 2,6-dioxopiperazines **1a–c** into the oxopiperazines **7a–c** and **8a–c** by using NaBH₄ in neat TFA. However, in contrast to the reactivity of **1a–c** toward NaBH₄ in MeOH, these dioxopiperazines were recovered unaltered. This different behavior depending on the solvent showed up the suggested participation of MeOH in the reduction of carbonyl groups by NaBH₄.¹³

Taking into account the potential tautomerization of the *N*-acyliminium intermediate **4** to the enamine **5**, and thereby the possible racemization,¹⁴ the enantiomeric purity of the reaction products was investigated. Firstly, we attempted this objective by means of chiral HPLC, using a CHIRALPAK IA column (150×4.6 mm, 5 μm) for the analysis of the reduction products. For this purpose, the reactions shown in **Scheme 2** were repeated starting from the enantiomeric (*R*)-form of the 2,6-dioxopiperazines **1a–c**, derived from the corresponding D-amino acid derivative. Then, all the reaction products were analyzed on chiral HPLC, and the results for each enantiomeric pair were compared. Unfortunately, only the phenylalanine-derived pair **7a** showed enough enantiomeric resolution to be able to quantify racemization (ee=76). As an alternative methodology, the racemization was indirectly determined by the measure of D-incorporation at position 5, replacing NaBH₄ with NaBD₄ and using deuterated solvents (CD₃OD and deuterated TFA) in the two steps of **Scheme 2**. Then, the D-incorporation was determined by the integral decrease for 5-H in the ¹H NMR spectra. As the racemization should be due to the hydride addition to the enamine intermediate **5**, the percentage of D-incorporation at C₅ should be twice the percentage of racemization. This study showed the absence of racemization in the first step of addition of hydride to the 2,6-dioxopiperazines **1a–c**, while a variable rate of racemization was determined for the reduction of the 6-hydroxy-2-oxopiperazines **2a–c** (ee=80 for **7a**, 60 for **7b**, and 0 for **7c**).

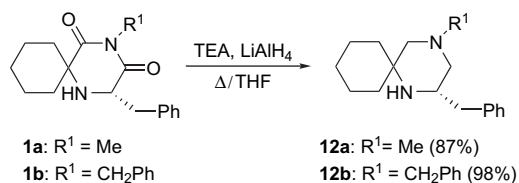
As an alternative for the regioselective synthesis of the 6-oxopiperazine **8a**, we studied its preparation by reduction of the L-phenylalanine-derived α-amino nitrile **9**¹⁰ (**Scheme 3**). Initially, the reduction of the cyano group was unsuccessfully attempted by catalytic hydrogenation. Thus, using 10% Pd(C) or Raney Ni as catalyst at room temperature and 1–3 atm of H₂ pressure, conditions previously used for the reduction of other α-amino-nitriles,¹⁵ the starting material was recovered unaltered after 2 days. Similarly, the attempts of cyano-reduction by hydrogen transfer from hydrazine monohydrate¹⁶ or formate¹⁷ were also unsuccessful. Finally, the cyano-reduction was achieved using NaBH₄ in the presence of CoCl₂.¹⁸ As shown in **Scheme 3**, the treatment of **9** with 10 equiv of NaBH₄ and 2 equiv of CoCl₂ in MeOH led to a 42% of the *N*-unsubstituted-6-oxopiperazine **10**, along with a 21% of the *N*-cyclohexyl-L-phenylalanine derivative

11, resulting from the removal of HCN, followed by reduction of the intermediate imine. Alkylation of **10** by treatment with MeI in the presence of Cs₂CO₃ gave a 90% of **8a**.



Scheme 3. Regioselective synthesis of 6-oxopiperazine derivatives.

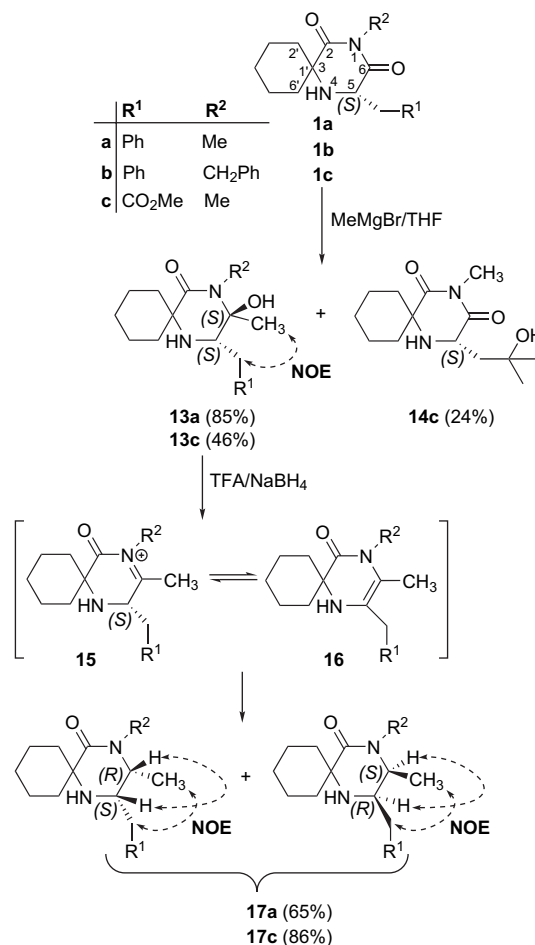
Having in mind the recently reported regioselective reduction of glutarimide derivatives to δ -lactams by treatment with Et₃N, followed by reaction with LiAlH₄ in refluxing THF,¹⁹ we also studied the application of these reaction conditions to the L-phenylalanine-derived dioxopiperazines **1a,b**. In this case, LiAlH₄ reduced both carbonyl groups at C₂ and C₆ to give the respective piperazines **12a** and **12b** (Scheme 4).



Scheme 4. LiAlH₄ mediated reduction of 2,6-dioxopiperazines.

In parallel to the reduction studies, the reductive alkylation of 2,6-dioxopiperazines **1a–c** was studied via reaction with Grignard reagents, such as MeMgBr, EtMgBr, and PhCH₂MgCl. As shown in Scheme 5, the reactivity of 1-methyl-2,6-dioxopiperazines **1a** and **1c** toward MeMgBr was, in part, similar to that commented above for their reduction with NaBH₄ in MeOH, however, the 1-benzyl analog **1b** was unreactive. The addition of MeMgBr to **1a** and **1c** was completely regioselective at C₆, and, as in the case of the reduction, also completely diastereoselective for the *syn* addition with respect to the amino acid side chain, to give the 6-hydroxy-2-oxopiperazine derivatives **13a** and **13c**, respectively. In the case of the aspartic acid derivative **1c**, the alcohol **14c**, resulting from a double addition of MeMgBr to the methyl ester of the side chain, was obtained as side product. The assignment of the nucleophilic addition position was based on the ¹H–¹³C correlations observed in the HMBC spectra of **13a** and **13c**. Particularly, those of C₂ with 2'-H or 6'-H, and 5-H with C₆ and 6-CH₃. In a second reaction step, the treatment of the 6-hydroxy-2-oxopiperazines **13a** and **13c** with NaBH₄ in neat TFA led, via the

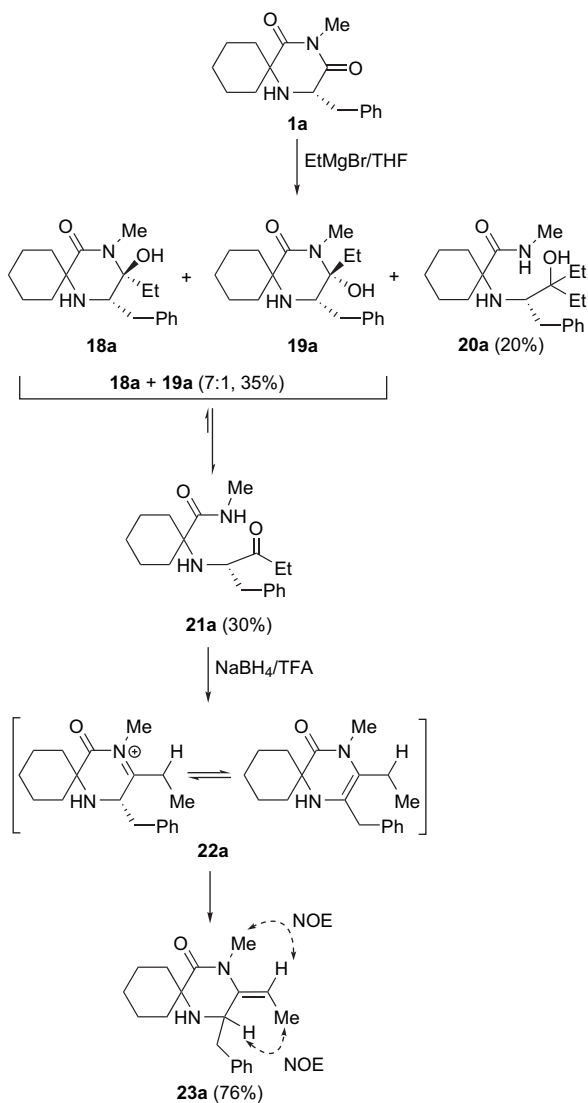
intermediates **15** and **16**, to the 2-oxopiperazines **17a** and **17b**. The assignment of relative configuration at C₆ was based on the NOE effects indicated in Scheme 5. According to this assignment, and contrary to the commented addition of NaBH₄ and MeMgBr to 2,6-dioxopiperazines, the addition of hydride to the *N*-acyliminium ions was mainly governed by steric factors, thus, the nucleophile attacked the intermediate **15** from the less hindered face, *anti* to the amino acid side chain.



Scheme 5. Reaction of 2,6-dioxopiperazines **1a–c** with MeMgBr.

Steric factors were also responsible for the different reactivity of these spirocyclic 2,6-dioxopiperazines with Grignard reagents more voluminous than MeMgBr. Thus, as shown in Scheme 6, the reaction of **1a** with EtMgBr gave a complex mixture, which was chromatographically resolved into the (\approx 7:1) diastereoisomeric mixture of 6-hydroxy-2-oxopiperazines **18a** and **19a** (35%), their ketone tautomer **21a** (30%), and the tertiary alcohol **20a** (20%), resulting from the attack of EtMgBr to **21a**. Interestingly, the diastereoisomeric mixture of **18a** and **19a** was unstable due to their transformation into the ketone **21a**. In view of this transformation,^{8c} the second step of NaBH₄ mediated reduction was carried out with the **18a+19a+21a** mixture, which gave the 6-ethylidene-2-oxopiperazine derivative **23a** (76%) as the only reaction product. The NOE effects observed in the ¹H 1D NOESY spectrum of **23a**, between the N–Me and the olefinic proton, and between 5-H and the allylic protons, allowed the assignment of *E* configuration to the double bond. The preference

for the deprotonation to the reduction reaction due to the slight increase in volume from the Me to the Et group showed that the entrance of hydride was strongly limited by the high steric crowding in the *N*-acyliminium ion. Similarly, the steric hindrance to the entrance of the nucleophile could be the reason for the unreactivity of the 2,6-dioxopiperazines **1a–c** toward PhCH_2MgCl .



Scheme 6. Reaction of the 2,6-dioxopiperazine **1a** with EtMgBr .

The racemization rate in the two steps of the reductive alkylations of Schemes 5 and 6 was also determined by chiral HPLC and by the measure of D-incorporation at C₅. Both methodologies showed a complete racemization (*ee*=0) for the second reaction step in the synthesis of **17a**, **17c**, and **23a**.

3. Conclusion

In conclusion, the overall results herein disclosed show the potential of spirocyclic 2,6-dioxopiperazine as a source of diverse piperazine derivatives, via regio- and diastereoselective reduction and reductive alkylation. The addition of NaBH_4 or Grignard reagents to the 2,6-dioxopiperazine

ring is mainly governed by electronic factors, which determine the preferential entrance of the nucleophile to C₆ from the most hindered face, *syn* to the amino acid side chain. However, in the resulting 6-hydroxypiperazines, the diastereoselectivity of the acid-mediated dehydration/ NaBH_4 addition is determined by steric factors, to give the product of nucleophilic attack from the less hindered face, *anti* to the amino acid side chain. This second reduction step produces partial or total racemization. The steric crowding of the 3-spiro-2,6-dioxopiperazine skeleton limits its reactivity to hydride and low volume Grignard reagents.

4. Experimental

4.1. General

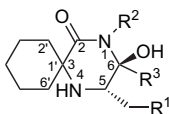
All reagents were of commercial quality. Solvents were dried and purified by standard methods. Analytical TLC was performed on aluminum sheets coated with a 0.2 mm layer of silica gel 60 F₂₅₄. Silica gel 60 (230–400 mesh) was used for flash chromatography. Preparative circular chromatography was performed on 20 cm diameter glass plates coated with a 1-mm layer of silica gel PF₂₅₄. Analytical RP-HPLC was performed on a Novapak C₁₈ (3.9 × 150 mm, 4 μm) column, with a flow rate of 1 mL/min, and using a tunable UV detector set at 214 nm. Mixtures of CH_3CN (solvent A) and 0.05% TFA in H_2O (solvent B) were used as mobile phases. Analytical chiral HPLC was carried out on a CHIRALPAK IA (150 × 4.6 mm, 5 μm) column of Daicel Chemical Industries Ltd., with a flow rate of 1 mL/min. EtOH of 10–1% in hexane was tried as mobile phase, although the best enantiomer resolution was obtained with 1% of EtOH. ¹H NMR spectra were recorded at 300 or 400 MHz, using TMS as reference, and ¹³C NMR spectra were recorded at 75 or 100 MHz. The NMR spectral assignment was based on COSY, HSQC, and HMBC spectra. ES-MS spectra were performed, in positive mode, using MeOH as solvent.

4.2. General procedure for the NaBH_4 mediated reduction of 2,6-dioxopiperazine-3-spirocyclohexanes **1a–c**. Synthesis of the hydroxypiperazines **2a–c** and **3a–c**

NaBH_4 (28.4 mg, 0.75 mmol) was slowly added to a 0 °C cooled solution of the corresponding 2,6-dioxopiperazines **1a–c**¹⁰ (0.25 mmol) in MeOH (4 mL). After reaching room temperature, the reaction mixture was stirred for 2 h. Then, the solvent was removed under reduced pressure and the residue was dissolved in CH_2Cl_2 (20 mL). This solution was successively washed with H_2O (5 mL) and brine (5 mL), dried over Na_2SO_4 , and evaporated to dryness. The residue was purified by flash chromatography, using 6–35% gradient of MeOH in CH_2Cl_2 as eluant. In this purification the 6-hydroxy-2-oxopiperazines **2a–c**, of higher *R_f*, were separated from their regioisomers **3a–c**. Significant analytical and spectroscopic data of these hydroxypiperazines are summarized in Tables 1 and 2.

4.3. General procedure for the NaBH_4 mediated reduction of hydroxypiperazines. Synthesis of the oxopiperazines **7a–c**, **8a**, and **17a,c**

NaBH_4 (23.8 mg, 0.63 mmol) was slowly added to a 0 °C cooled solution of the corresponding hydroxypiperazines

Table 1. Significant analytical and spectroscopic data of 6-hydroxy-2-oxopiperazines

	2a	2b	2c	13a	13c	18a+19a
R ¹	Ph	Ph	CO ₂ Me	Ph	CO ₂ Me	Ph
R ²	Me	CH ₂ Ph	Me	Me	Me	Me
R ³	H	H	H	Me	Me	Et
Formula ^a	C ₁₇ H ₂₄ N ₂ O ₂	C ₂₃ H ₂₈ N ₂ O ₂	C ₁₃ H ₂₂ N ₂ O ₄	C ₁₈ H ₂₆ N ₂ O ₂	C ₁₄ H ₂₄ N ₂ O ₄	C ₁₉ H ₂₈ N ₂ O ₂
Yield (%)	85	87	82	85	46	35
ES-MS [M+1] ⁺	289.2	365.1	271.2	303.2	285.1	317.1
Mp (°C)	169–171 (Benzene)	156–157 (Benzene)	Syrup	Syrup	Syrup	Syrup
t _R (min) (A:B) ^b	2.31 (30:70)	3.84 (40:60)	1.59 (25:75)	2.11 (50:50)	1.89 (25:75)	7.79 (14%) and 8.39 (86%) (25:75)
¹ H NMR ^c						
5-H	2.94	3.00	3.21	2.92	3.23	3.09, 3.23
5-CH ₂	2.51, 3.25	2.49, 3.15	2.52, 2.87	2.45, 3.20	2.34, 2.80	2.50, 2.56, 2.62
2'-H ^{ax}	1.94	2.09	2.09	1.95	2.04	1.10–1.98
Cyclohexyl	1.01–1.71	1.11–1.80	1.25–1.93	1.06–1.58	1.24–1.90	1.10–1.98
R ¹	7.23	7.14–7.33	3.72	7.30	3.72	7.13–7.77
R ²	2.85	4.38, 5.11, 7.14–7.33	2.92	2.91	2.89	2.89, 2.92
R ³	4.46	4.49	4.46	1.46	1.32	0.90, 1.04, 1.87, 1.91
J _{5,6} (Hz)	7.5	7.5	7.5	—	—	—
OH	ND ^d	2.74	ND ^d	ND ^d	2.49	ND ^d
¹³ C NMR ^c						
C ₂	175.0	174.5	174.9	174.2	174.5	ND ^d
C ₃	58.1	58.1	58.3	58.6	58.9	ND ^d
C ₅	55.7	56.1	51.9	60.0	56.0	ND ^d
C ₆	84.4	81.9	83.8	86.6	85.9	ND ^d
5-CH ₂	38.3	38.2	36.7	35.9	35.2	ND ^d
R ¹	126.9, 128.7, 129.0, 137.8	126.6, 128.4, 128.7, 137.2	52.0, 172.4	127.0, 128.7, 128.9, 138.5	52.0, 173.1	ND ^d
R ²	30.1	44.84, 127.0, 128.6, 129.1 137.5	30.1	26.8	26.7	ND ^d
R ³	—	—	—	19.1	18.9	ND ^d
Cyclohexyl	19.8, 20.6, 24.8, 29.4, 35.6	19.7, 20.5, 24.6, 30.1, 35.7	20.3, 20.7, 25.1, 30.1, 35.6	19.7, 20.7, 24.7, 29.1, 36.2	20.5, 20.9, 25.1, 30.3, 36.3	ND ^d

^a Satisfactory analysis for C, H, and N.

^b Novapak C₁₈ (3.9×150 mm, 4 μm). A=CH₃CN, B=0.05% TFA in H₂O.

^c Spectra registered at 300 or 400 MHz, in CDCl₃, assigned with the help of COSY spectra.

^d ND=not determined.

^e Spectra registered at 75 or 100 MHz, in CDCl₃, assigned with the help of HSQC and HMBC spectra.

2a–c, **3a**, and **13a,c** (0.21 mmol) in TFA (3 mL). After reaching room temperature, the reaction mixture was stirred for 3 h. Then, the solvent was removed under reduced pressure and the residue was dissolved in CH₂Cl₂ (20 mL). This solution was successively washed with H₂O (5 mL) and brine (5 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by flash chromatography, using 10–40% gradient of MeOH in CH₂Cl₂ as eluant. The significant analytical and spectroscopic data of the 6-oxopiperazine **8a** are summarized in Table 2, while those of the 2-oxopiperazines **7a–c** and **17a,c** are summarized in Table 3.

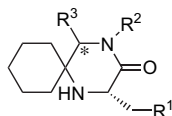
4.4. Synthesis of (5S)-5-phenylmethyl-6-oxopiperazine-3-spirocyclohexane (**10**) and N-cyclohexyl-L-phenylalanine methyl ester (**11**)

NaBH₄ (171.7 mg, 4.5 mmol) and CoCl₂ (118 mg, 0.9 mmol) were slowly added to a 0 °C cooled solution of N-(1-cyanocyclohexyl)-L-phenylalanine methyl ester (**9**)¹⁰ (128.7 mg, 0.45 mmol) in MeOH (7 mL). After reaching

room temperature, the reaction mixture was stirred for 1 h. Then, the solvent was removed under reduced pressure and the residue was dissolved in EtOAc (20 mL). This solution was successively washed with H₂O (5 mL) and brine (5 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by flash chromatography, using 10–50% gradient of EtOAc in hexane as eluant. In this purification the 6-oxopiperazine **10** (49.2 mg, 42%), whose significant analytical and spectroscopic data are summarized in Table 2, was separated from N-cyclohexyl-L-phenylalanine methyl ester (**11**).

4.4.1. N-Cyclohexyl-L-phenylalanine methyl ester (**11**).

Syrup (25.7 mg, 21%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 25:75)] t_R 5.46 min; ¹H NMR (300 MHz, CDCl₃) 1.00–1.76 (m, 10H, cyclohexyl), 2.27 (m, 1H, 1'-H), 2.79 (dd, 1H, J=7 and 13.5 Hz, 3-H), 2.89 (dd, 1H, J=7 and 13.5 Hz, 3-H), 3.54 (s, 3H, OCH₃), 3.58 (t, 1H, J=7 Hz, 2-H), 7.17 (m, 5H, Ph); ES-MS m/z 262.0 [M+1]⁺. Anal. Calcd for C₁₆H₂₃NO₂: C, 73.53; H, 8.87; N, 5.36. Found: C, 73.74; H, 8.92; N, 5.16.

Table 2. Significant analytical and spectroscopic data of 6-oxopiperazines

	3a	3b	3c	8a	10
R ¹	Ph	Ph	CO ₂ Me	Ph	Ph
R ²	Me	CH ₂ Ph	Me	Me	H
R ³	OH	OH	OH	H	H
Formula ^a	C ₁₇ H ₂₄ N ₂ O ₂	C ₂₃ H ₂₈ N ₂ O ₂	C ₁₃ H ₂₂ N ₂ O ₄	C ₁₇ H ₂₄ N ₂ O	C ₁₆ H ₂₂ N ₂ O
Yield (%)	15	13	18	100	42
ES-MS [M+1] ⁺	289.2	365.2	271.2	273.2	259.3
t _R (min) (A:B) ^b	2.32 (30:70)	3.50 (40:60)	2.77 (25:75)	2.83 (25:75)	2.06 (25:75)
¹ H NMR ^c					
2-H	4.42	4.40	4.44	3.02, 3.07	2.94, 3.04
5-H	3.66	3.80	3.88	3.52	3.63
5-CH ₂	2.67, 3.52	2.75, 3.59	2.57, 3.11	2.77, 3.28	2.88, 3.29
Cyclohexyl	1.11–1.67	1.02–1.62	1.15–1.62	1.29–1.70	1.19–1.64
R ¹	7.27	7.12–7.38	3.70	7.27	7.21
R ²	3.01	4.22, 5.22, 7.12–7.38	3.00	2.91	6.12
¹³ C NMR ^d					
C ₂	86.3	83.1	87.0	60.7	51.9
C ₃	54.0	58.4	55.6	50.7	49.6
C ₅	56.0	56.5	51.0	56.2	54.6
C ₆	170.5	170.3	173.0	170.3	171.8
5-CH ₂	39.9	40.3	38.7	39.9	38.0
R ¹	126.6, 128.7, 129.2, 138.6	126.7, 128.5, 129.4, 137.4	51.6, 172.6	127.3, 129.3, 130.2, 140.6	126.6, 128.5, 129.2, 138.1
R ²	32.1	47.3, 127.5, 128.7, 129.4, 137.6	33.8	36.7	—
Cyclohexyl	20.9, 21.6, 25.6, 29.8, 34.4	21.2, 21.3, 25.5, 31.8, 34.6	20.8, 21.2, 25.7, 29.9, 33.8	22.5, 27.1, 32.1, 38.1	21.5, 21.6, 25.9, 30.6, 37.1

^a Syrups, satisfactory analysis for C, H, and N.

^b Novapak C₁₈ (3.9×150 mm, 4 μm). A=CH₃CN, B=0.05% TFA in H₂O.

^c Spectra registered at 300 or 400 MHz, in CDCl₃, assigned with the help of COSY spectra.

^d Spectra registered at 75 or 100 MHz, in CDCl₃, assigned with the help of HSQC and HMBC spectra.

4.5. Regioselective synthesis of (5S)-1-methyl-5-phenylmethyl-6-oxopiperazine-3-spirocyclohexane (8a)

MeI (16.3 μL, 0.26 mmol) and Cs₂CO₃ (85.2 mg, 0.26 mmol) were successively added under argon to a solution of the 6-oxopiperazine **10** in dry CH₃CN (2.5 mL). This reaction mixture was stirred at 50 °C for 2 h. Then, the solvent was evaporated and the residue was dissolved in CH₂Cl₂ (20 mL). The solution was successively washed with H₂O (5 mL) and brine (5 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by circular chromatography, using 15–25% gradient of EtOAc in hexane as eluant, to obtain the 1-methyl-6-oxopiperazine **8a** (43 mg, 90%).

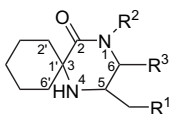
4.6. General procedure for the synthesis of piperazines 12a and 12b

TEA (25.1 μL, 0.18 mmol) was added under argon to a solution of the corresponding 2,6-dioxopiperazines **1a** and **1b** (0.15 mmol) in dry THF (4 mL). After 30 min of stirring at room temperature, LiAlH₄ (17.1 mg, 0.45 mmol) was added to the reaction mixture, and it was heated at reflux temperature for 3 h. Then, the reaction mixture was cooled at room temperature and saturated solution of NH₄Cl (1 mL) was added to destroy the excess of LiAlH₄. The precipitated Al(OH)₃ was filtered off and washed with THF.

The combined filtrates were evaporated to dryness and the residue was dissolved in EtOAc (20 mL). This solution was successively washed with H₂O (5 mL) and brine (5 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by flash chromatography, using 20–45% gradient of EtOAc in hexane as eluant.

4.6.1. (5S)-1-Methyl-5-phenylmethylpiperazine-3-spirocyclohexane (12a). Syrup (33.7 mg, 87%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 25:75)] t_R 1.79 min; ¹H NMR (300 MHz, CDCl₃) δ 1.08–1.57 (m, 10H, cyclohexyl), 1.62 (t, 1H, J=10.5 Hz, 6-H), 1.64 (d, 1H, J=11 Hz, 2-H), 2.18 (s, 3H, 1-CH₃), 2.55 (dd, 1H, J=8 and 13 Hz, 5-CH₂), 2.64 (dd, 1H, J=6 and 13 Hz, 5-CH₂), 2.67 (d, 1H, J=11 Hz, 2-H), 2.73 (dd, 1H, J=2.5 and 10.5 Hz, 6-H), 3.26 (m, 1H, J=2.5, 6, and 8 Hz, 5-H), 7.23 (m, 5H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 21.6 and 21.8 (C_{3'} and C_{5'}), 26.2 (C_{4'}), 31.8 and 38.5 (C_{2'} and C_{6'}), 41.0 (5-CH₂), 46.6 (CH₃), 50.2 (C₅), 51.4 (C₃), 62.1 (C₆), 64.7 (C₂), 126.3, 128.4, 129.0, and 138.5 (Ph); ES-MS m/z 259.3 [M+1]⁺. Anal. Calcd for C₁₇H₂₆N₂: C, 79.02; H, 10.14; N, 10.84. Found: C, 79.22; H, 10.24; N, 10.72.

4.6.2. (5S)-1,5-Di(phenylmethyl)piperazine-3-spirocyclohexane (12b). Syrup (50.1 mg, 98%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 40:60)] t_R 8.42 min; ¹H NMR (300 MHz, CDCl₃) δ 1.04–1.53 (m, 10H, cyclohexyl), 1.59

Table 3. Significant analytical and spectroscopic data of 2-oxopiperazines

	7a	7b	7c	17a	17c
R ¹	Ph	Ph	CO ₂ Me	Ph	CO ₂ Me
R ²	Me	CH ₂ Ph	Me	Me	Me
R ³	H	H	H	Me	Me
Formula ^a	C ₁₇ H ₂₄ N ₂ O	C ₂₃ H ₂₈ N ₂ O	C ₁₃ H ₂₂ N ₂ O ₃	C ₁₈ H ₂₆ N ₂ O	C ₁₄ H ₂₄ N ₂ O ₃
Yield (%)	100	100	100	70	90
ES-MS [M+1] ⁺	273.2	349.2	255.1	287.3	269.2
t _R (min) (A:B) ^b	3.47 (25:75)	17.20 (25:75)	1.64 (25:75)	1.78 (50:50)	1.82 (25:75)
¹ H NMR ^c					
5-H	4.38	3.80	4.35	3.34	3.63
6-H	3.28, 3.79	3.18, 3.50	3.58, 3.89	3.15	3.22
5-CH ₂	3.06, 3.56	2.94, 3.18	2.98, 3.24	2.58, 2.69	2.37, 2.46
2'-H ^{ax}	2.28	2.23	2.19–2.30	1.97	2.05
Cyclohexyl	1.49–2.24	1.34–2.09	1.35–2.30	1.10–1.65	1.15–1.87
R ¹	7.32	7.22–7.33	3.73	7.22–7.33	3.71
R ²	2.85	4.38, 5.11, 7.14–7.33	2.92	2.91	2.89
R ³	—	—	—	1.23	1.16
J _{5,6} (Hz)	4.5 and 11.5	4 and 11.5	4.5 and 11.5	3.5	3.5
¹³ C NMR ^d					
C ₂	167.1	172.3	172.6	174.0	172.4
C ₃	62.2	61.6	63.7	58.8	59.9
C ₅	49.7	50.6	47.6	52.4	48.7
C ₆	49.9	51.4	50.7	58.2	57.7
5-CH ₂	36.4	239.2	34.5	38.3	36.2
R ¹	127.7, 129.1, 129.5, 135.1	128.7, 130.0, 130.6, 138.7	53.4, 168.4	126.7, 128.7, 128.8, 138.2	52.6, 172.4
R ²	32.9	51.4, 128.7, 130.0, 130.6, 138.7	35.7	34.2	31.3
R ³	—	—	—	12.6	13.1
Cyclohexyl	20.4, 20.8, 24.3, 31.1, 34.4	21.6, 22.2, 26.3 32.4, 35.7	21.3, 21.5, 25.6, 31.9, 33.7	20.2, 20.9, 24.9, 30.3, 36.2	20.9, 21.2, 25.4, 30.1, 34.7

^a Satisfactory analysis for C, H, and N.

^b Novapak C₁₈ (3.9×150 mm, 4 μm). A=CH₃CN, B=0.05% TFA in H₂O.

^c Spectra registered at 300 or 400 MHz, in CDCl₃, assigned with the help of COSY spectra.

^d Spectra registered at 75 or 100 MHz, in CDCl₃, assigned with the help of HSQC and HMBC spectra.

(d, 1H, *J*=11 Hz, 2-H), 1.74 (t, 1H, *J*=10.5 Hz, 6-H), 2.47 (dd, 1H, *J*=8.5 and 13.5 Hz, 5-CH₂), 2.58 (d, 1H, *J*=11 Hz, 2-H), 2.60 (dd, 1H, *J*=5.5 and 13.5 Hz, 5-CH₂), 2.75 (dd, 1H, *J*=2.5 and 10.5 Hz, 6-H), 3.21 (m, 1H, *J*=2.5, 5.5, and 8.5 Hz, 5-H), 3.25 (d, 1H, *J*=13.5 Hz, 1-CH₂), 3.50 (d, 1H, *J*=13.5 Hz, 1-CH₂), 7.11–7.29 (m, 10H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 21.6 and 21.7 (C_{3'} and C_{5'}), 26.2 (C_{4'}), 31.8 and 38.4 (C_{2'} and C_{6'}), 41.0 (5-CH₂), 50.5 (C₅), 51.6 (C₃), 60.6 (C₆), 61.9 (C₂), 62.9 (1-CH₂), 126.3, 126.8, 128.1, 128.4, 128.6, 129.0, 138.8, and 138.9 (Ph); ES-MS *m/z* 335.2 [M+1]⁺. Anal. Calcd for C₂₃H₃₀N₂: C, 82.59; H, 9.04; N, 8.37. Found C, 82.43; H, 9.24; N, 8.57.

4.7. General procedure for the reaction of 2,6-dioxopiperazines 1a–c with Grignard reagents. Synthesis of hydroxypiperazines 13a,c, 18a, and 19a

The corresponding Grignard reagent, MeMgBr, EtMgBr (3 M solution in diethyl ether, 200 μL, 0.6 mmol), or PhCH₂MgCl (1 M solution in diethyl ether, 600 μL, 0.6 mmol), was added under argon to a –40 °C cooled solution of the 2,6-dioxopiperazines 1a–c (0.2 mmol) in dry THF (3 mL), and the reaction mixture was stirred at this

temperature for 3 h. Afterward, the reaction mixture was evaporated to dryness and the residue was dissolved in CH₂Cl₂ (20 mL). The solution was successively washed with H₂O (5 mL) and brine (5 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by circular chromatography, using 15–40% gradient of EtOAc in hexane as eluant, to give the compounds shown in Schemes 5 and 6. Significant analytical and spectroscopic data of the 6-hydroxy-2-oxopiperazines 13a, 13c, and 18a+19a are summarized in Table 1.

4.7.1. (5S)-5-(2-Hydroxy-2-methyl)propyl-1-methyl-2,6-dioxopiperazine-3-spirocyclohexane (14c). Syrup (12.9 mg, 24%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 25:75)] t_R 4.03 min; ¹H NMR (300 MHz, CDCl₃) δ 1.31 and 1.33 [2s, 6H, Me (isopropyl)], 1.25–1.62 (m, 8H, cyclohexyl), 1.69 (dd, 1H, *J*=10.5 and 14.5 Hz, 5-CH₂), 2.03 (m, 1H, 2'-H^{ec}), 2.21 (m, 1H, 6'-H^{ax}), 2.44 (dd, 1H, *J*=2.5 and 14.5 Hz, 5-CH₂), 3.12 (s, 3H, 1-Me), 3.86 (dd, 1H, *J*=2.5 and 10.5 Hz, 5-H); ¹³C NMR (75 MHz, CDCl₃) δ 20.5 (C_{3'} and C_{5'}), 24.3 (C_{4'}), 26.7 (1-Me), 28.2 and 31.1 [Me (isopropyl)], 29.2 and 34.6 (C_{2'} and C_{6'}), 42.2 (5-CH₂), 51.3 (C₅), 58.9 (C₃), 70.5 (isopropyl), 173.1 (C₆), 176.1 (C₂); ES-MS *m/z* 269.2 [M+1]⁺.

Anal. Calcd for C₁₄H₂₄N₂O₃: C, 62.66; H, 9.01; N, 10.44. Found: C, 62.46; H, 8.79; N, 10.62.

4.7.2. N-Methyl-1-[(2S)-3-ethyl-3-hydroxy-1-phenylpentan-2-yl]aminocyclohexylcarboxamide (20a). Syrup (24.2 mg, 20%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 25:75)] *t*_R 9.95 min; ¹H NMR (300 MHz, CDCl₃) δ 0.70 and 0.90 [2t, 6H, *J*=7.5 Hz, CH₃ (Et)], 0.95–1.82 (m, 10H, cyclohexyl), 1.45 and 1.49 [2q, 4H, CH₂ (Et)], 2.61 (dd, 1H, *J*=7.5 and 14.5 Hz, CH₂-Ph), 2.68 (dd, 1H, *J*=6.5 and 14.5 Hz, CH₂-Ph), 2.74 (d, 3H, *J*=5 Hz, NH-CH₃), 3.17 (dd, 1H, *J*=6.5 and 7.5 Hz, NH-CH), 6.37 (br s, 1H, NH-CH₃), 7.23 (m, 5H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ 7.2 and 7.3 [CH₃ (Et)], 22.0, 22.1, 25.4, 33.2, 35.2, and 60.5 (cyclohexyl), 26.3 (NH-CH₃), 28.3 and 28.6 [CH₂ (Et)], 38.6 (CH₂-Ph), 57.6 (NH-CH), 74.7 (C-OH), 126.1, 128.3, 129.1, and 140.1 (Ph), 177.3 (CONH); ES-MS *m/z* 347.1 [M+1]⁺. Anal. Calcd for C₂₁H₃₄N₂O₂: C, 72.79; H, 9.89; N, 8.08. Found: C, 73.12; H, 10.04; N, 7.96.

4.7.3. N-Methyl-1-[(2S)-3-oxo-1-phenylpentan-2-yl]aminocyclohexylcarboxamide (21a). Syrup (33.2 mg, 30%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 25:75)] *t*_R 17.27 min; ¹H NMR (300 MHz, CDCl₃) δ 0.97 [(t, 3H, *J*=7.5 Hz, CH₃ (Et)], 1.02–1.92 (m, 10H, cyclohexyl), 2.18 (d, 3H, *J*=5 Hz, NH-CH₃), 2.36 (dd, 1H, *J*=9.5 and 13.5 Hz, CH₂-Ph), 2.39 [q, 2H, *J*=7.5 Hz, CH₂ (Et)], 2.71 (dd, 1H, *J*=4.5 and 13.5 Hz, CH₂-Ph), 3.29 (dd, 1H, *J*=4.5 and 9.5 Hz, NH-CH), 6.05 (br s, 1H, NH-CH₃), 7.22 (m, 5H, Ph); ¹³C NMR (75 MHz, CDCl₃) δ (ppm): 7.5 [CH₃ (Et)], 21.6, 22.0, 25.2, 31.8, 36.8, and 61.5 (cyclohexyl), 25.8 (NH-CH₃), 35.3 [CH₂ (Et)], 41.9 (CH₂-Ph), 63.6 (NH-CH), 126.9, 128.6, 129.6, and 138.0 (Ph), 176.7 (CONH), 214.9 (COEt); ES-MS *m/z* 317.1 [M+1]⁺. Anal. Calcd for C₁₉H₂₈N₂O₂: C, 72.12; H, 8.92; N, 8.85. Found: C, 72.22; H, 8.75; N, 9.11.

4.8. Synthesis of 6-[(E)-ethylidene-1-methyl-5-phenylmethyl-2-oxopiperazine-3-spirocyclohexane (23a)

NaBH₄ (18.7 mg, 0.49 mmol) was slowly added to a 0 °C cooled solution of a mixture of the 6-hydroxypiperazines **18a+19a** and their ketone epimer **21a** (52.0 mg, 0.16 mmol) in TFA (3 mL). After reaching room temperature, the reaction mixture was stirred for 3 h. Then, the solvent was removed under reduced pressure and the residue was dissolved in CH₂Cl₂ (20 mL). This solution was successively washed with H₂O (5 mL) and brine (5 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by flash chromatography, using 10–40% gradient of MeOH in CH₂Cl₂ as eluant, to give **23a** (37.2 mg, 76%); HPLC [Novapak C₁₈ (3.9×150 mm, 4 μm) (A:B, 25:75)] *t*_R 9.01 min; ¹H NMR (500 MHz, CDCl₃) δ 1.24–1.65 (m, 9H, cyclohexyl), 1.51 [d, 3H, *J*=7 Hz, CH₃ (ethylidene)], 2.03 (m, 1H, 2'-H^{ax}), 2.87 (dd, 1H, *J*=6 and 13.5 Hz, 5-CH₂), 2.91 (dd, 1H, *J*=6 and 13.5 Hz, 5-CH₂), 2.98 (s, 3H, 1-CH₃), 4.19 (t, 1H, *J*=6 Hz, 5-H), 4.96 [q, 1H, *J*=7 Hz, CH (ethylidene)], 7.23 (m, 5H, Ph); ¹³C NMR (125 MHz, CDCl₃) δ 12.0 [CH₃ (ethylidene)], 21.1, 21.3, 25.6 (C₅), 31.6 and 33.7 (cyclohexyl), 32.0 (1-CH₃), 40.6 (5-CH₂), 51.4 (C₅), 57.0 (C₃), 104.6 [CH (ethylidene)], 126.7, 128.3, 129.4, and 137.5 (Ph), 139.7 (C₆), 175.0 (C₂);

ES-MS *m/z* 299.3 [M+1]⁺. Anal. Calcd for C₁₉H₂₆N₂O: C, 76.47; H, 8.78; N, 9.39. Found: C, 76.59; H, 8.48; N, 9.17.

4.9. Determination of the racemization ratio by the measure of deuterium incorporation

The procedures described above for the synthesis of oxopiperazines **7a–c**, **8a**, **17a**, **17c**, and **23a** were carried out replacing NaBH₄ by NaBD₄ and the corresponding solvents MeOH and TFA by CD₃OD and deuterated TFA. The resulting products were chromatographically (TLC and HPLC) identical to the respective nondeuterated compounds, as well as their ¹H NMR data, except for the commented decrease in the signals corresponding to the protons, which have been partial or completely replaced by deuterium. The 50% decrease in the integral of 5-H was used as a measure for the percentage of configuration inversion at C₅ for the calculation of the enantiomer excesses.

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References and notes

- For some recent reviews on privileged structures, see: (a) Patchett, A. A.; Nargund, R. P. *Annual Reports in Medicinal Chemistry*; Doherty, A. M., Ed.; Academic: San Diego, CA, 2000; Vol. 35, pp 289–298; (b) Klabunde, T.; Hessler, G. *ChemBioChem* **2002**, *3*, 928–944; (c) Horton, D. A.; Bourne, G. T.; Smythe, M. L. *J. Comput. Aided Mol. Des.* **2002**, *16*, 415–430; (d) Horton, D. A.; Bourne, G. T.; Smythe, M. L. *Chem. Rev.* **2003**, *103*, 893–930; (e) Bondensgaard, K.; Ankersen, M.; Thøgersen, H.; Hansen, B. S.; Wulff, B. S.; Bywater, R. P. *J. Med. Chem.* **2004**, *47*, 888–899.
- For some reviews on 2,5-dioxopiperazines, see: (a) Witiak, D. T.; Wei, Y. *Progress in Drug Research*; Jucker, E., Ed.; Birkhäuser: Basel, 1990; Vol. 35, pp 249–363; (b) Prasad, C. *Peptides* **1995**, *16*, 151–164; (c) Dinsmore, C. J.; Beshore, D. C. *Tetrahedron* **2002**, *58*, 3297–3312; (d) Fischer, P. M. *J. Pept. Sci.* **2003**, *9*, 9–35.
- (a) Andoh, T.; Ishida, R. *Biochim. Biophys. Acta* **1998**, *1400*, 155–171; (b) Andoh, T. *Biochimie* **1998**, *80*, 235–246; (c) Nair, V. N.; Witiak, D. T. *J. Chem. Educ.* **1998**, *65*, 534–538.
- (a) Tomassini, J. E.; Davies, M. E.; Hastings, J. C.; Lingham, R.; Mojena, M.; Raghoobar, S. L.; Singh, S. B.; Tkacz, J. S.; Goetz, M. A. *Antimicrob. Agents Chemother.* **1996**, *40*, 1189–1193; (b) Singh, S. B.; Tomassini, J. E. *J. Org. Chem.* **2001**, *66*, 5504–5516.
- (a) Challis, B. C.; Challis, J. A. *Comprehensive Organic Chemistry. Synthesis and Reactions of Organic Compounds*; Barton, D., Ollis, W. D., Eds.; Pergamon: Oxford, 1979; Vol. 2, pp 957–1065; (b) Hiemstra, H.; Speckamp, W. N. *Comprehensive Organic Synthesis. Selectivity, Strategy and Efficiency in Modern Organic Chemistry*; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 2, pp 1047–1082.

6. (a) Speckamp, W. N.; Hiemstra, H. *Tetrahedron* **1985**, *41*, 4367–4416; (b) Speckamp, W. N.; Moolenaar, M. J. *Tetrahedron* **2000**, *56*, 3817–3856; (c) Maryanoff, B. E.; Zhang, H.-C.; Cohen, J. H.; Turchi, I. J.; Maryanoff, C. A. *Chem. Rev.* **2004**, *104*, 1431–1628.
7. (a) Ryu, Y.; Kim, G. *J. Org. Chem.* **1995**, *60*, 103–108; (b) Pilli, R. A.; Russowsky, D. *J. Org. Chem.* **1996**, *61*, 3187–3190; (c) Ostendorf, M.; Romagnoli, R.; Pereiro, I. C.; Roos, E. C.; Moolenaar, M. J.; Speckamp, W. N.; Hiemstra, H. *Tetrahedron: Asymmetry* **1997**, *8*, 1773–1789; (d) Hsu, R.-T.; Cheng, L.-M.; Chang, N.-C.; Tai, H.-M. *J. Org. Chem.* **2002**, *67*, 5044–5047; (e) Gill, C. D.; Greenhalgh, D. A.; Simpkins, N. S. *Tetrahedron* **2003**, *59*, 9213–9230; (f) Simpkins, N. S.; Gill, C. D. *Org. Lett.* **2003**, *5*, 535–537.
8. (a) Schuch, C. M.; Pilli, R. A. *Tetrahedron: Asymmetry* **2000**, *11*, 753–764; (b) Schuch, C. M.; Pilli, R. A. *Tetrahedron: Asymmetry* **2002**, *13*, 1973–1980; (c) Liu, L.-X.; Ruan, Y.-P.; Guo, Z.-Q.; Huang, P.-Q. *J. Org. Chem.* **2004**, *69*, 6001–6009; (d) Chen, M.-C.; He, M.-Z.; Zhou, X.; Huang, L.-Q.; Ruan, Y.-P.; Huang, P.-Q. *Tetrahedron* **2005**, *61*, 1335–1344.
9. Dixon, R. A.; Jones, S. *Tetrahedron: Asymmetry* **2002**, *13*, 1115–1119.
10. González-Vera, J. A.; García-López, M. T.; Herranz, R. *J. Org. Chem.* **2005**, *70*, 3660–3666.
11. (a) Cieplak, A. S. *J. Am. Chem. Soc.* **1981**, *103*, 4540–4552; (b) Danishefsky, S. J.; Cabal, M. P.; Chow, K. *J. Am. Chem. Soc.* **1989**, *111*, 3456–3457; (c) Jeroncic, L. O.; Cabal, M.-P.; Danishefsky, S. J. *J. Org. Chem.* **1991**, *56*, 387–395.
12. (a) Gutiérrez-Rodríguez, M.; García-López, M. T.; Herranz, R. *Tetrahedron* **2004**, *60*, 5177–5183; (b) Alonso De Diego, S. A.; Gutiérrez-Rodríguez, M.; Pérez de Vega, M. J.; González-Muñiz, R.; Herranz, R.; Martín-Martínez, M.; Cenarruzabeitia, E.; Frechilla, D.; Del Río, J.; Jimeno, M. L.; García-López, M. T. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 3396–3400.
13. Smith, M. B.; March, J. *March's Advanced Organic Chemistry. Reactions, Mechanisms, and Structure*, 5th ed.; Wiley-Interscience: New York, NY, 2001; pp 1197–1203.
14. (a) Beausoleil, E.; L'Archeveque, B.; Belec, L.; Atfani, M.; Lubell, W. D. *J. Org. Chem.* **1996**, *61*, 9447–9454; (b) Martín-Martínez, M.; Bartolome-Nebreda, J. M.; Gómez-Monterrey, I.; González-Muñiz, R.; García-López, M. T.; Ballaz, S.; Barber, A.; Fortuño, A.; Del Río, J.; Herranz, R. *J. Med. Chem.* **1997**, *40*, 3402–3407.
15. Suárez-Gea, M. L.; García-López, M. T.; González-Muñiz, R.; Herrero, S.; Herranz, R. *Tetrahedron Lett.* **1996**, *37*, 2083–2084.
16. (a) Herrero, S.; García-López, M. T.; Herranz, R. *J. Org. Chem.* **2003**, *68*, 4582–4585; (b) Herrero, S.; García-López, M. T.; Cenarruzabeitia, E.; Del Río, J.; Herranz, R. *Tetrahedron* **2003**, *59*, 4491–4499.
17. Gowda, S.; Gowda, C. *Tetrahedron* **2002**, *58*, 2211–2213.
18. (a) Osby, J. O.; Heinzman, S. W.; Ganem, B. *J. Am. Chem. Soc.* **1986**, *108*, 67–72; (b) Gaucher, A.; Bintein, F.; Wakselman, M.; Mazaleyrat, J. P. *Tetrahedron Lett.* **1998**, *39*, 575–578.
19. Chen, C.-Y.; Chang, B.-R.; Tsai, M.-R.; Chang, M.-Y.; Chang, N.-C. *Tetrahedron* **2003**, *59*, 9383–9387.